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320 Chapter 10 • The Mole Section 110.10.1 Measuring Matter
MAIN Idea Chemists use the mole to count atoms, molecules, ions, and formula units. Real-World Reading Link Has your class ever had a contest to guess how

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Mole 1.. Idenü\$-' and calculaterthe number of.representa- five particles. in each of the following quantities. ... Chemistry: Matter and Change Chapter 11 13 . CHAPTER 12. Determine the empirical formula for a 100.00-g sample of a compound having the following percent composition.

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The MoleThe Mole

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Likewise, a mole will always contain 6.022×10^{23} particles. Although a mole of any substance is the same number, different substances have different molar masses. The molar mass of a compound can be calculated from its chemical formula and can be used to convert from mass to moles of that compound.

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