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answer for each multiple choice. For all equilibrium reactions the arrow will be double-headed (\leftrightarrow). Remember to balance any chemical reactions before calculating if appropriate. 1.

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the end of the exam, the proctors and I collect everything from the students. As soon as the exam is over, I go to my office, give the multiple choice part of the exam back to the students with an answer key.

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The exam has 7 pages plus the Periodic Table and Reference page.

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Key 3 ... you will have
a maximum of 2.5
hours to complete your
final exam. 4 Grade 11
Chemistry. Part a: Fill-
in-the-Blanks (22
marks) ... 8 Grade 11
Chemistry. 10. In which
of the following is
concentration
expressed in percent

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by volume? a) 10%
(v/v)

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CHEM PRINCIPLES 1
- BU**

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Final Exam A Summer
2015 - Confidential
Exam ID: 70 Course
Name: - Current Date
and Time: Mon, Aug
10, 2015 @ 14:58:27
EDT Total Exam Points:
26.00 Question #: 1 If
a solid line represents

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a covalent bond and a dotted line represents an intermolecular

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PART I - Multiple choice
(3 points each). Do
NOT mark on the test.
Write your name and

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answers in Scantron. 1.
Round off the following
measurement to the
four significant digits:
107.300 g A) 107 g B)
107.0 g C) 107.3 g D)
107.5 g 2.

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A Summer 2015 -
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Identify the letter of
the choice that best
completes the
statement or answers
the question. ____ 1.
The study of chemicals
that, in general, do not
contain carbon is
traditionally called
what type of

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chemistry? a. bio c.
physical b. inorganic d.
analytical ____ 2. Which
of the following is NOT
an example of matter?

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3. Dehydration.

Oxidation. reduction.

Neutralization. ...

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Valence Shell Electron

Pair Repulsion.

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A—Answers 1. D 2. B 3.
E 4. C 5. D 6. A 7. C 8.
B 9. C 10. D 11. A 12. E
13.

**Final Practice
examination answer
Key - Manitoba**

Chem 107 PRACTICE
Page 16/24

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EXAM II Page 4 14. (8 pts) A hydrocarbon is composed of 88.8 % carbon and 11.2 % hydrogen. a) What is its empirical formula? Show your work clearly and write your final answer on the

Chem 107 PRACTICE EXAM II CCBC- Catonsville

While not to the same extent as Group I metals, Group II metals are quite reactive with

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water. Each metal has a different level of reactivity with water, but they ultimately follow a well-observed reactivity trend.

PART I Multiple choice (3 points each). - HCC

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Chem 107 Final Exam
There are 50 multiple choices that are worth 3 points each. There are 3 problems and 1 bonus problem. Try to

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answer the questions, which you know first, and then solve the problems, which you are not sure about. Here is the summary of the total 200 points plus 10 bonus points. Carefully read the questions. Good luck!

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Version # A On the last page of this exam, you've been given a periodic table and some physical constants.

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CHEMISTRY 121 FINAL EXAM Record your answers on the machine readable answer sheet provided. Scratch work is to be done on the paper

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provided with the
examination
instructions. DO NOT
MAKE ANY MARKS ON
THIS EXAMINATION
BOOK. QUESTIONS
1-50 COUNT ONE
POINT EACH. 100

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Chemistry 107 is
designed as the first
semester of a two-
semester sequence in
introductory chemistry.

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keys, etc.

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