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CHAPTER 7 REVIEW

1 Core Instruction Instruction and Intervention Support Chemical Formulas and Chemical Compounds Chapter Resources The Teacher's Edition wrap has extensive teaching support for every lesson, including Misconception Alerts, Teach from Visuals, Demonstrations,

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Teaching Tips, Differentiated Instruction,
and Problem Solving support.

Chemical Formulas and Chemical Chapter 7 Compounds

CHAPTER 7 REVIEW Chemical Formulas
and Chemical Compounds SECTION 1
SHORT ANSWER 1. In a Stock system
name such as iron(III) sulfate, the

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Roman numeral tells us (a) (b) (c) (d) 2.
Answer the following questions in the
space provided.

7 Chemical Formulas and Chemical Compounds

Modern Chemistry 57 Chemical Bonding
CHAPTER 7 REVIEW Chemical Formulas
and Chemical Compounds SECTION 3

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SHORT ANSWER Answer the following questions in the space provided. 1. Label each of the following statements as True or False: _____ a. If the formula mass of one molecule is x u, the molar mass is x g/mol. _____ b.

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Chapter 7: Chemical Formulas and
Chemical Compounds Section 2:
Oxidation Numbers Oxidation numbers
An oxidation number, also called
oxidation state, indicates the general
distribution of electrons among the
bonded atoms in a molecular compound
or polyatomic ion. Unlike ionic charges,
oxidation numbers do not have an exact

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physical meaning.

7 Chemical Formulas and Chemical Compounds

CHAPTER 7 REVIEW Chemical Formulas
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REVIEW SHORT ANSWER Answer the
following questions in the space
provided. 1. Write formulas for the

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following compounds: CuCO₃ a.
copper(II) carbonate Na₂SO₃ b. sodium
sulfite (NH₄)₃PO₄ c. ammonium
phosphate SnS₂ d. tin(IV) sulfide HNO₂
e. nitrous acid

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and Chemical Compounds SECTION 3
SHORT ANSWER Answer the following
questions in the space provided. 1. Label
each of the following statements as True
or False: True a. If the formula mass of
one molecule is x amu, the molar mass
is x g/mol. False b. Samples of equal
numbers of moles of two different
chemicals

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Chapter 7: Chemical Formulas and Chemical Compounds

Chapter 7 - Chemical Formulas & Chemical Compounds Chapter 7 is the final chapter of the first semester. It serves as a summary of all course content discussed so far, as the focus of this chapter is on process skills rather

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than new concepts or theory.

**Holt McDougal Modern Chemistry
Chapter 7: Chemical ...**

Chemical Formulas and Chemical
Compounds. MIXED REVIEW. ... Its molar
mass is 128.18 g/mol and it contains
93.75% carbon and 6.25% hydrogen.
Determine the molecular formula of

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naphthalene from this information. Name
Class Date Name Class Date ... CHAPTER
7 REVIEW ...

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chemical formulas. In this chapter, you
will be introduced to some of the rules
used to identify simple chemical

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compounds. Significance of a Chemical Formula Recall that a chemical formula indicates the relative number of atoms of each kind in a chemical compound. For a molecular compound, the

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and Chemical Compounds SECTION 2
SHORT ANSWER Answer the following
questions in the space provided. 1.
Assign the oxidation number to the
specified element in each of the
following examples: a. S in H_2SO_3 b. S in
 MgSO_4 c. S in K_2S d. Cu in Cu_2S e. Cr in
 Na_2CrO_4 f. N in HNO_3 g. C in (HCO_3) h.
N in (NH_4) 2. a.

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1 Core Instruction

Chapter 7 Section 1 Chemical Names and Formulas Significance of a Chemical Formula, continued •The chemical formula for an ionic compound represents one formula unit—the simplest ratio of the compound's positive ions (cations) and its negative

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ions (anions). • example: aluminum sulfate —Al₂(SO₄)₃

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Chemical Formulas and Chemical

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the following

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compounds chemical formulas ...**

Chapter 6 "Chemical Bonding" Chapter 7
"Chemical Formulas and Chemical
Compounds" Chapter 8 "Writing and
Balancing Equations" Chapter 9

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"Stoichiometry" Chapter 11 "Gas Laws"
Chapter 14 "Lewis Dot Structures, Acids
& Bases" VSEPR Chapters 12 & 15
"Molarity, pH & Titrations" Chapter 18
"Chemical Equilibrium"

**Chapter 7 "Chemical Formulas and
Chemical ... - Chemistry**

7. What is the empirical formula for

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lactic acid, which has a molecular formula of $C_3H_5O_3$. a. CH_2O b. CHO c. CHO d. CHO

8. $NaCl$ is a. a molecular formula only. b. an empirical formula only. c. both an empirical formula and a chemical formula. d. both a molecular formula and an empirical formula.

9.

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Modern Chemistry 61 Chemical Bonding
CHAPTER 7 REVIEW Chemical Formulas
and Chemical Compounds MIXED
REVIEW SHORT ANSWER Answer the
following questions in the space
provided. 1. Write formulas for the
following compounds: _____ a. copper(II)

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carbonate _____ b. sodium sulfite _____ c.
ammonium phosphate

**chapter 7 - I 7 Q CHAPTER 7 REVIEW
Chemical Formulas and ...**

Chapter 7 - Chemical Formulas and
Chemical Compounds 7-1 Chemical
Names and Formulas I. Significance of a
Chemical Formula A. Molecular formulas

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1. Number of atoms of each element in one molecule of a compound C_2H_6 = ethane (2 carbon atoms, 6 hydrogen atoms) B. Ionic Compounds 1.

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following

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questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on ...

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